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Short communication

Simple stationary phases derived from gluconolactone for chiral high-performance liquid chromatography

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Abstract

A new family of chiral stationary phases (CSPs) have been prepared by chemical modification of $\text{p-}\delta$ -gluconolactone with ring opening. They were chemically bonded to $5-\mu\text{m}$ microporous silica and evaluated as column packing materials for chiral analysis by HPLC. The best was the CSP in which the hydroxyl groups derived from the ring-opened gluconolactone were converted to carbamate residues using a naphthylethyl isocyanate.

1. Introduction

Lactones have been used in the preparation of chiral stationary phases (CSPs) [1] for HPLC and as chiral analytes [2] in GC. These applications demonstrate their ability to take part in chiral recognition processes with either an open chain or a closed ring system. With the earlier work of Lourenco [3] in mind, a number of novel CSPs for HPLC were synthesised from a cheap and readily available carbohydrate, D-δ-gluconolactone (I).

2. Experimental

2.1. Instrumentation

The HPLC system consisted of a Shimadzu LC-5A single-head pump, Rheodyne 7125 manual injector valve and Cecil Instruments CE 212 variable-wavelength UV monitor (254 nm).

2.2. Materials

HPLC-grade solvents were obtained from Rathburn. N-3,5-Dinitrobenzoyl (DNB) derivatives of various amino acids (Aldrich and Janssen) were synthesised at UMIST using established methods [4]. Hypersil silica (5 μ m) was supplied by Shandon HPLC.

2.3. Preparation of CSPs

CSPs 1-5 were prepared according to Fig. 1. The synthesis involved nucleophilic ring opening of the lactone (I) with allylamine, followed by acetylation and conversion to an active silane

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Fig. 1. Synthesis of novel CSP from gluconolactone. RO = Aryl-NH-(CO)-O and HO, according to Table 1. Reagents: A = allylamine; $B = Ac_2O$; $C = (MeO)_3Si(CH_2)_3SH$; D = silica; $E = Me_3SiCl$; $F = NH_3 - MeOH$; G = aryl isocyanate.

derivative. The acetylated silane (II) was bonded to 5- μ m Hypersil silica. Residual silanols on the silica surface were then end-capped with trimethylsilyl (TMS) chloride. This was followed by deacetylation and carbamate formation. The CSPs were then packed into standard steel HPLC columns (15 cm \times 4.5 mm I.D.) by Shandon HPLC.

3. Results and discussion

The surface coverages of the phases were estimated from their % carbon contents. CSP 1

contained 237 μ mol of chiral selector and 422 μ mol of TMS groups per gram of silica. The corresponding values for CSP 2 were 274 and 174 μ mol g⁻¹. The chiral selector strands on CSPs 3–5 were not fully carbamated: the ratios of carbamate to hydroxyl groups on these phases are shown in Table 1.

Some reasonable separations were obtained on the acetylated phase, CSP 1. The chromatograms of a series of racemic N-3,5-DNB-amino acid methyl esters (III-VIII) on this phase are presented in Fig. 2. The hydroxyl-containing phase, CSP 2, showed very poor enantioselectivity.

The carbamate phases (CSPs 3-5) all afforded better enantioresolutions than CSP 1. Fig. 3 compares the α values for the same amino acid derivatives on the carbamate CSPs. As the carbamate is made more electron-rich from phenyl (CSP 3) to dimethylphenyl (CSP 4) to naphthylethyl (CSP 5), the separation of these electron-deficient analytes improves. This suggests that a π - π interaction is important in the chiral recognition process on these CSPs. Fig. 4 shows the resolution of the racemic samples (III-VIII) on the best performing phase, CSP 5.

Table 1 Structures of carbamate CSPs

Phase	OR group on strands of chiral selector	Concentration of strands	Concentration of TMS	OR:OH ratio on chiral strand
CSP 3	0 0-C	270	170	3.1:1.9
CSP 4	O-C-NH-Me	270	170	2.3:2.7
CSP 5	O C NH CH CH ₃	180	70	1.0:4.0

Concentrations are reported as μ mol per gram of silica gel. Average ratios of OR:OH groups on the chiral strands are also quoted.

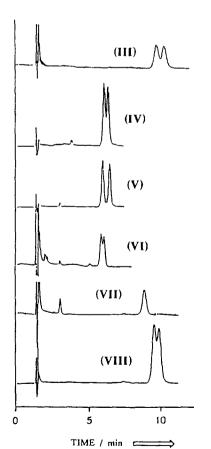


Fig. 2. Chromatograms of racemic N-3.5-DNB-amino acid methyl esters on CSP 1. Mobile phase: tetrahydrofuran (THF)-*n*-hexane (15:85) at 1.0 ml min⁻¹.

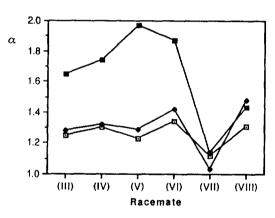


Fig. 3. Comparison of the α values for the racemic N-3,5-DNB-amino acid methyl esters on the carbamate phases CSP 3 (\square), CSP 4 (\spadesuit) and CSP 5 (\blacksquare). Mobile phase: THF-*n*-hexane (25:75) at 1.0 ml min⁻¹.

4. Conclusions

A family of simple CSPs for HPLC were prepared from a cheap and readily available starting material, $\text{D-}\delta\text{-gluconolactone}$. These chiral stationary phases were used to separate the enantiomers of amino acid derivatives that contained the electron-deficient 3,5-DNB group. The phase that shows the most commercial promise (CSP 5) contains the naphthylethyl carbamate functionality.

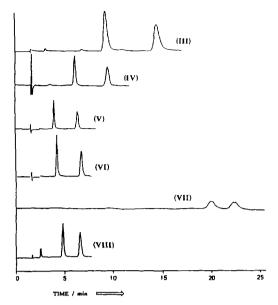


Fig. 4. Chromatograms of racemic N-3,5-DNB-amino acid methyl esters on CSP 5. Mobile phase as in Fig. 3.

Acknowledgements

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References

- [1] D. Lohmann and R. Dappen, Chirality, 5 (1993) 168.
- [2] S. Brochu et al., Polymer Bull., 30 (1993) 223.
- [3] W. Lourenco, Ph.D. Thesis, UMIST, Manchester, UK (1989).
- [4] W.H. Pirkle and C.J. Welch, J. Org. Chem., 49 (1984) 138.